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Temperature dependence of fracture toughness of sheet molybdenum was studied in the temperature range -196 °C...300 °C. It was found that nonmonotonous behaviour of this dependence is due to the change in crack nucleation mechanism.

## EXPERIMENTAL PROCEDURE

In the present study temperature dependence of fracture toughness of molybdenum sheet 2.25 mm wide of alloy CM-10 was investigated, samples were annealed at 2000 °C for one hour. Grain size was obtained about 300-400  $\mu$ m. Critical strength intensity factor was used as fracture toughness,  $K_{\rm LC}$  or  $K_{\rm C}$  were used as general yield has been arisen. Flat samples with two-edge notch with 0.05 mm radius been arisen. Flat samples with two-edge notch with 0.05 mm radius been subjected to tension at the rate  $10^{-3}$  s  $^{-1}$  in the temperature rangeof-196 °C...300 °C. Tensile tests at temperature higher than room temperature were performed in vacuum chamber ture higher than room temperature were performed in vacuum chamber with pressure of 0.1 Pa. Samples shape, size and method of fracture toughness calculation were chosen in accordance with Soviet standard (1). Initial cracks appeared at the tip of electrospark notch, their length being 20-50 Arm.

## RESULTS AND DISCUSSION

Temperature dependence of fracture toughness of annealed CM-10 alloy is shown in Figure 1 (curve 1). As may be seen from this fi

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gure, fracture toughness in this temperature range is nonmonotonous and exhibits maximum at room temperature. At subsequent temperature increase in fracture toughness is again observed in the range  $100\ ^{\rm O}{\rm C}\ldots300\ ^{\rm O}{\rm C}$ .

Analysis of mechanical tests shows that conditions of plane strain deformation are preserved in the temperature range up to  $70\,^{\circ}\text{C}$ . This fact has attracted our interest to the investigation of fracture toughness since the transition from plane strain state to plane-stress one is usually accompanied with increase in fracture toughness according to Krasovsky (2).

Temperature dependence of fracture stress is similar to that of fracture toughness (Fig. 1, curve 2). Fracture stress exceeds yield strength when the temperature exceeds room temperature and plasticity becomes noticeable. These two facts together with the fractographic peculiarities which will be described later allows us in accordance with Ioffe-Orovan model (3) to determine room temperature as a lower temperature limit of brittle-ductile transition of notched samples.

Fractographic analysis shows that in the temperature range -196 °C...20 °C fracture by cleavage proceeds from electrospark cracks, blunted as the test temperature is increased. Therefore, at liquid nitrogen temperature these cracks are hardly visible (Fig. 2a). They are clearly seen at room temperature (Fig. 2b). In some cases cleavage cracks can also start from favourably orientated grain boundaries.

In the temperature range above room temperature samples are also fractured by cleavage. Pure cleavage is preceded now by cleavage cracks with relaxation (Fig. 2c). These ones are according to Friedel-Orlov mechanism (4, 5) and propagates from blunted electrospark cracks. Length of cleavage cracks with relaxation increases as the test temperature is increasing Fig. 1 (curve 5). Stereoscopic analysis shows that cleavage cracks with relaxation are much sharper than blunted electrospark ones. Growth of introduced cracks to their critical length leads to the change in net specimen cross-section which is to be taken into account in calculation of fracture toughness. At the temperature 300 °C ductile fracture elements, intergranular and transgranular delamination(Fig. 2d) are also observed at the fracture surface together with the cleavage and cleavage with relaxation.

Thus, the increase of fracture toughness below temperature  $^{\rm T}_{\rm bd}$  results from blunting of electrospark cracks due to plastic processes at the crack tip. At  $^{\rm T}_{\rm bd}$  temperature microplastic blunting of these cracks is so strong that cleavage is not be able to nucleate.

As can be seen from Fig. l (curve 5), length of crack with relaxation increases from 0.015 mm at  $\rm T_{bd}$  to 0.052 mm at 100  $^{\rm o}{\rm C}.$ 

This fact must lead to increase of Klc. However level of fracture stress resulting from already nucleated crack sharpness and from structural changes in deformed material at the crack tip, which are difficult to evaluate today, is reduced from 370 to 220 MPa in the temperature range under discussion.

Fracture toughness as it is known may be calculated as:

as it is known may be
$$K_{1c} = 6_F \sqrt{\pi c}$$
(1)

Analysis of the influence of changes both of crack length and fracture stress on fracture toughness of molybdenum sheet clearly shows that in the observed temperature range the main reason for decrease toughness is reduction of fracture stress but not inctracture toughness is reduction of fracture stress but not the rease of a crack length, according to the dependence between fracture stress and crack length (1). Further rise of temperature with the stress and crack length (1) according to the continuous rise of fracture stress and crack length (1). continuos rise of fracture stress results in essential growth of the relaxation crack (Fig. 1, curves 2 and 5). This fact explains a the relaxation crack (Fig. 1, curves 2 and 3). Into fact experiments the relaxation crack (Fig. 1, curves 2 and 3). Into fact experiments ange, new increasing of fracture toughness in the temperature range over 70 °C in the temperature range over 70 °C. Increase of fracture toughness in the temperature range over 70 results also from the change in the specimen stress state - from results also from the change in the specimen stress state - from plane strain state to plane-stress one. As follows from Fig. 1 (curve 4), plasticity before fracture increases at these temperatu-(curve 4), plasticity before fracture increases at these temperatures. Size of the plastic zone is comparable to the specimen width. Therefore specimen becomes too "thin" for correct calculation of the specimen becomes too "thin" for correct calculation of the specimen specimen factor K<sub>1c</sub> according to formulae (1). The specimen that the specimen that the specimen will be specimen to the specimen with the specimen with the specimen will be specime ve l') and coefficient  $K_{\rm C}$  taking into account crack growth (Fig. 1, curve l'').

In some work, e.g., done by Vasilev et al (6), Lung and Gao (7) it has been shown that, using fractographic determination of fracture toughness of molybdenum, temperature dependence of fracture toughness of material failured by cleavage may be expressed by exponential relation:

$$K_{AC} = K_{o} \exp \left(-\frac{V_{o}}{kT}\right)$$
(2)
(3) is monotonous and differs from

It is seen, that dependence (2) is monotonous and differs from

As follows from data showed above, the main reason of the nondependence found in this work. monotony of temperature dependence of fracture toughness of molybdenum results from the change of fracture mechanisms and, more exactly, from the change of mechanisms of bearing of cleavage. Be-To the cleavage cracks are originated on blunted electrospark The cleavage cracks are originated on blunted electrospank the cleavage cracks are originated on cracks of cracks, after Tbd, the cleavage cracks are originated on cracks of cleavage with plastic relaxation. The sharpness of cracks of cleavage with plastic relaxation, than blunted electrospark ones. vage with relaxation is more high, than blunted electrospark ones.

Because the mechanism of full fracture in the both cases is the same, i. e. cleavage, it is possible to describe the found temperature dependence of fracture toughness of molybdenum by two exponents, activation energies of which are equal, but differing by some constants appropriate for different temperature intervals: before  $T_{\rm bd}$  and after it.

Scheme of the temperature dependence of fracture toughness of molybdenum is represented by Figure 3. The transition from the first exponent to the second one happens when cleavage cracks are preceded by cracks of cleavage with relaxation.

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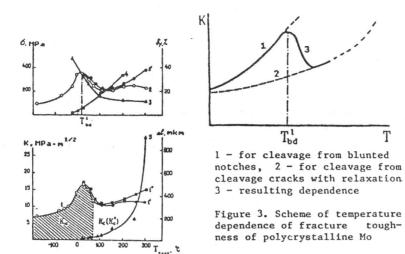


Figure 1. Temperature dependence of mechanical properties of polycrystalline Mo. (1, 1', 1'' - fracture toughness, 2, 2' - fracture strength, 4 - plasticity before fracture, 5 - length of cleavage crack with relaxation)

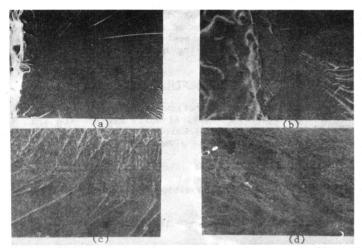


Figure 2. SEM  $\,$  view of the fracture of polycrystalline Mo at different temperatures